

**Spring 2024**

**MECH 2311 (23178) INTRODUCTION TO THERMAL-FLUID SCIENCE**

**Course Description**      An introduction to basic concepts of thermodynamics and fluid mechanics to include properties, property relationships, states, and fluids. Presentation of the basic equations of thermal-fluid science, continuity, first and second laws of thermodynamics, and momentum.  
*Prerequisites are MATH 1312 Calculus II with a grade of "C" or better.*

**Instructor**      Miguel Cedeno, Ph.D.  
Assistant Professor of Instruction  
3D Lockheed Martin Lab Administrator  
Rocket Team Director  
Aerospace and Mechanical Engineering

[https://outlook.office365.com/owa/calendar/OfficeHoursSpring2024@minersutep.onmicro  
soft.com/bookings/](https://outlook.office365.com/owa/calendar/OfficeHoursSpring2024@minersutep.onmicrosoft.com/bookings/)

*E-mail address:*      macedenomor@utep.edu  
*Office:*                  Engineering Building Room E-330  
*Office hours:*        Friday via MS Teams 8-9 am  
*Office Phone:*        915-747-7976

[Click here for Booking](#)

**Additional Tutoring and Grades Review**      Teaching Assistant:  
Office Hours:

**Textbook**              *Fundamentals of Thermal-Fluid Sciences* Sixth Edition (2021)  
by Yunus Cengel, John Cimbala, and Robert Turner  
McGraw Hill

**Course Content**      1. Introduction the Thermodynamics and Fluid Mechanics (Chapter 1)

Part 1

2. Basic Concepts of Thermodynamics (Chapter 2)
  - a. Systems and Control Volumes
  - b. Properties of a System
  - c. Density and Specific Gravity
  - d. State and Equilibrium
  - e. Zeroth Law of Thermodynamics
  - f. Pressure
  - g. The Manometer

3. Energy, Energy Transfer, and General Energy Analysis (Chapter 3)

- a. Forms of Energy
  - b. Energy Transfer by Heat
  - c. Energy Transfer by Work
  - d. Mechanical Forms of Work
  - e. First Law of Thermodynamics
  - f. Energy Conversion Efficiencies
4. Properties of Pure Substances (Chapter 4)
- a. Phases of a Pure Substance
  - b. Phase-Change Processes of Pure Substances
  - c. Property Diagrams for Phase-Change Processes
  - d. Property Tables
  - e. The Ideal-Gas Equation
5. Energy Analysis of Closed Systems (Chapter 5)
- a. Moving Boundary Work
  - b. Energy Balance for Closed Systems
  - c. Specific Heats
  - d. Internal Energy, Enthalpy, and Specific Heats of Ideal Gases
  - e. Internal Energy, Enthalpy, and Specific Heats of Liquids
6. Mass and Energy Analysis of Control Volumes (Chapter 6)
- a. Conservation of Mass
  - b. Flow Work and the Energy of a Flowing Fluid
  - c. Energy Analysis of Steady-Flow Systems
7. The Second Law of Thermodynamics (Chapter 7)
- a. Introduction to the second law
  - b. Thermal Energy Reservoirs
  - c. Heat Engines
  - d. Refrigerators and Pumps
  - e. Reversible and Irreversible Processes
  - f. The Carnot Cycle
  - g. The Carnot Principles
  - h. Thermodynamic Temperature Scale
  - i. Carnot Heat Engine
  - j. The Carnot Refrigerator and Pump

## Part 2

8. Fluid Statics (Chapter 11)
- a. Introduction to Fluid Statics
  - b. Hydrostatic Forces on Submerged Plane Surfaces
9. Bernoulli and Energy Equations (Chapter 12)

- a. The Bernoulli Equation
- b. General Energy Equation
- c. Energy Analysis of Steady Flows

10. Momentum Analysis of Flow Systems (Chapter 13)

- a. Newton's Laws
- b. Choosing a Control Volume
- c. Forces Acting on a Control Volume
- d. The Reynolds Transport Theorem
- e. The Linear Momentum Equation

<b>Grading</b>	Quizzes	200
	Tests (4 given – 3 best will count)	300
	<b>Possible Points</b>	<b>500</b>

**A (500-450): B (449-400): C (399-350): D (349-300): F (299 and below)**

There will be ten 12-min-quizzes of 20 points each and four 80-minute tests of 100 points each. There will be no makeup for the tests or quizzes. Your lowest test grade will be dropped.

**Optional HW Extra Credits:** At Sunday midnight of each week, HW will be due. Submission is optional, but strongly recommended. Solving HW would be beneficial to reinforce concepts covered in class. At the end of the semester, the total extra credit worth of HW, would be calculated based on the performance of the course. To get the max allowed extra credits, it is recommended to submit every HW in a weekly based, since the total points would be proportional to the number of submissions. Once the submission link expires at the end of each week, it will not be activated again.

**Major Course Objectives**

Upon completion of this course, students should be able to:

1. Understand concepts of temperature, pressure, and energy
2. Evaluate properties of pure substances and use property data for solving problems
3. Apply the principles of conservation of mass and energy to closed and open systems
4. Understand and apply the second law of thermodynamics, including concepts such as irreversibility and Carnot cycle
5. Understand the basic concepts of fluid mechanics and properties such as viscosity and surface tension
6. Solve fluid statics problems
7. Apply the Bernoulli and energy equations
8. Apply the conservation of linear momentum to control volumes

**Policy for Quizzes**

Quizzes will be based on helping you study for both the Fundamental of Engineering (FE) and Professional Engineering (PE) exams

[\(http://www.ncees.org/exams/\)](http://www.ncees.org/exams/).

I recommend that you use an approved calculator like what you will use for your certification exams for all your work since this will help you learn how to use all the features of your calculator. These calculators include Casio: All fx-115 models. Any Casio calculator must contain fx-115 in its model name, Hewlett Packard: The HP 33s and HP 35s models, and Texas Instruments: All TI-30X and TI-36X models.

**ABET Program Outcomes Impacted** This class significantly addresses the following ABET objectives:  
(a) an ability to apply knowledge of mathematics, science, and engineering  
(b) an ability to design and conduct experiments, as well as to analyze and interpret data  
(e) an ability to identify, formulate, and solve engineering problems

### **Technology requirements**

Course content is delivered via the Internet through the Blackboard Collaborate ULTRA. Ensure your UTEP e-mail account is working and that you have access to the Web and a stable web browser. Google Chrome and Mozilla Firefox are the best browsers for Blackboard; other browsers may cause complications. When having technical difficulties, update your browser, clear your cache, or try switching to another browser.

You will need to have access to a computer/laptop, scanner, a webcam, and a microphone. You will need to download or update the following software: Microsoft Office, Adobe Acrobat Reader, Windows Media Player, QuickTime. Check that your computer hardware and software are up-to-date and able to access all parts of the course.

If you do not have a word-processing software, you can download Word and other Microsoft Office programs (including Excel, PowerPoint, Outlook and more) for free via UTEP's Microsoft Office Portal. Click the following link for more information about [Microsoft Office 365](#) and follow the instructions.

**IMPORTANT:** If you encounter technical difficulties beyond your scope of troubleshooting, please contact the UTEP [Help Desk](#) as they are trained specifically in assisting with technological needs of students. Please do not contact me for this type of assistance.

### **Course Communication:** *How we will stay in contact with each other*

There are several ways we can keep the communication channels open:

- **Office Hours:** My office hours will be held on MS Teams on Fridays 8-9am. If you need to meet with me at another time, feel free to stop by my office on campus when I'm there.
- **Email:** UTEP e-mail is the best way to contact me. I will make every attempt to respond to your e-mail within 24-48 hours of receipt. **When e-mailing me, be sure to email from your UTEP student ID and please put the course number in the subject line.** In the

body of your e-mail, clearly state your question. At the end of your e-mail, be sure to put your first and last name, and your UTEP ID.

- **Announcements:** Check the Blackboard announcements frequently for any updates, deadlines, or other important messages.

## **NETIQUETTE**

As we know, sometimes communication online can be challenging. It's possible to miscommunicate what we mean or to misunderstand what our classmates mean given the lack of body language and immediate feedback. Therefore, please keep these netiquette (network etiquette) guidelines in mind. Failure to observe them may result in disciplinary action.

- Always consider the audience. This is a college-level course; therefore, all communication should reflect polite consideration of other's ideas.
- Respect and courtesy must always be provided to classmates and to the instructor. No harassment or inappropriate postings will be tolerated.
- When reacting to someone else's message, address the ideas, not the person. Post only what anyone would comfortably state in a face-to-face situation.
- **Blackboard is not a public internet venue;** all postings to it should be considered private and confidential. Whatever is posted on in these online spaces is intended for classmates and professor only. Please do not copy documents and paste them to a publicly accessible website, blog, or other space such as Chegg.

**Course Policies:** *What do you need to do to be successful in the course?*

### **Attendance and participation**

Attendance on the course is determined by participation in the learning activities of the course. Your participation in the course is important not only for your learning and success but also to create a community of learners. Participation is determined by completion of the following activities:

- Reading/Viewing all course materials to ensure understanding of assignment requirements
- Other activities as indicated in the weekly modules such as HW (Self-study problems)
- Meet regularly with the TA during additional study sessions posted on BB

### **Academic dishonesty**

Academic dishonesty is prohibited and is considered a violation of the UTEP Handbook of Operating Procedures. It includes, but is not limited to, cheating, plagiarism, and collusion. Cheating may involve copying from or providing information to another student, possessing unauthorized materials during a test (Chegg), or falsifying research data on laboratory reports. Plagiarism occurs when someone intentionally or knowingly represents the words or ideas of another as one's own. Collusion involves collaborating with another person to commit any academically dishonest act. Any act of academic dishonesty attempted by a UTEP student is unacceptable and will not be tolerated. All suspected violations of academic integrity at The University of Texas at El Paso must be reported to the Office of Student Conduct and Conflict Resolution (OSCCR) for possible disciplinary action. To learn more, please visit [HOOP: Student Conduct and Discipline](#).

**Reasonable Accommodation Policy:** The University is committed to providing reasonable accommodations and auxiliary services to students, staff, faculty, job applicants, applicants for

admissions, and other beneficiaries of University programs, services and activities with documented disabilities in order to provide them with equal opportunities to participate in programs, services, and activities in compliance with sections 503 and 504 of the Rehabilitation Act of 1973, as amended, and the Americans with Disabilities Act (ADA) of 1990 and the Americans with Disabilities Act Amendments Act (ADAAA) of 2008. Reasonable accommodation will be made unless it is determined that doing so would cause undue hardship on the University. Students requesting an accommodation based on a disability must register with the [UTEP Center for Accommodations and Support Services](#) (CASS). Contact the Center for Accommodations and Support Services at 915-747-5148, or email them at [cass@utep.edu](mailto:cass@utep.edu), or apply for accommodations online via the [CASS portal](#).

### **Quizzes proctoring software**

Quizzes/Tests will make use of Respondus Lock Down Browser and Respondus Monitor inside of Blackboard to promote academic integrity. You are encouraged to learn more about how to use these programs prior to the first test. We're using Zoom Meetings to proctor the Tests, so have it ready in your phones.

Please review the following guidelines:

- The assessments will only be available at the times identified on the course calendar.
- You may take the test at a given time window.
- A reliable Internet connection is essential to completing the exam. If you must go to a location to take the exam (such as the library), be sure to follow their health and safety requirements.
- Respondus Lockdown Browser will require that all internet tabs are closed prior to the start of the test/quiz.
- Respondus Monitor requires a webcam and microphone.
- You will be required to show your webcam your student ID prior to the start of the test.
- Your face should be completely visible during the test. Blocking the camera will disable the test.
- No notes or textbook materials are permitted during the test. Respondus Monitor requires you to take a video of your surrounding area (desk, chair, walls, etc.)
- You should not have conversations with other people and/or leave and return to the area during the test.

### **Plagiarism detecting software**

Some of your course work and assessments may be submitted to SafeAssign, a plagiarism detecting software. SafeAssign is used to review assignment submissions for originality and will help you learn how to properly attribute sources rather than paraphrase.

### **Copyright statement for course materials**

All materials used in this course are protected by copyright law. The course materials are only for the use of students currently enrolled in this course and only for the purpose of this course. They may not be further disseminated.

**Course Resources:** Where you can go for assistance UTEP provides a variety of student services

and support:

#### Technology Resources

- [Help Desk](#): Students experiencing technological challenges (email, Blackboard, software, etc.) can submit a ticket to the UTEP Helpdesk for assistance. Contact the Helpdesk via phone, email, chat, website, or in person if on campus.

#### Academic Resources

- [UTEP Library](#): Access a wide range of resources including online full-text access to thousands of journals and eBooks plus reference service and librarian assistance for enrolled students.
- [University Writing Center \(UWC\)](#): Submit papers here for assistance with writing style and formatting, ask a tutor for help and explore other writing resources.
- [Math Tutoring Center \(MaRCS\)](#): Ask a tutor for help and explore other available math resources.
- [RefWorks](#): A bibliographic citation tool; check out the RefWorks tutorial and Fact Sheet and Quick-Start Guide.

#### Individual Resources

- [Military Student Success Center](#): Assists personnel in any branch of service to reach their educational goals.
- [Center for Accommodations and Support Services](#): Assists students with ADA-related accommodations for coursework, housing, and internships.
- [Counseling and Psychological Services](#): Provides a variety of counseling services including individual, couples, and group sessions as well as career and disability assessments.

### **Department of Aerospace and Mechanical Engineering Safety Statement**

The Department of Aerospace and Mechanical Engineering at the University of Texas at El Paso is committed to a model of excellence in education that includes providing a safe and healthy environment for its students, staff, faculty, and the public.

Our goal is to maximize education and research training that can only occur if you, the individual, minimize hazards and risks. This can be done by:

- Providing adequate control of the health and safety risks arising from all activities.
- Consulting with employees on matters affecting their health and safety
- Providing and maintaining safe laboratories and equipment.
- Ensuring safe handling and use of substance.
- Ensuring all employees are competent to do their task and have adequate training; and
- Maintaining clean, safe, and healthy working conditions

The principal investigator or individual in charge of each laboratory is ultimately responsible for safety in that respective lab. This includes training and ultimate release of the laboratory. Within the Department, we hold every employee (staff, faculty, student) responsible for implementing our safety practices and our departmental safety policy. We hold every employee (staff, faculty, student) responsible for providing leadership within our department to establish effective environmental safety and occupational health standards.

**CLASS SCHEDULE**  
**MEETING TIME: M W 130-250pm**  
**MEETING LOCATION: Classroom Building C305**

Week	Date	Chapter(s)	Read Sections	HW Optional (Textbook)	Scheduled and DUE
1	Monday, January 15, 2024	Dr. Martin Luther King, Jr. Holiday – University Closed			
	Wednesday, January 17, 2024	1 – Intro	1.1-6	-	-
2	Monday, January 22, 2024	2 – Basic Concepts of Thermodynamics	2.1-3	2-16C, 2-27C, 2-32E, 2-37	-
	Wednesday, January 24, 2024	-	2.4-8	2-38E, 2-49, 2-46, 2-52E	<b>Quiz 1 Covering Chap. 2</b>
3	Monday, January 29, 2024	3- Energy, Energy Transfer, and General Energy Analysis	3.1-5	3-10, 3-12, 3-13, 3-15, 3-17C, 3-22C, 3-27E, 3-28E, 3-30E, 3-35	-
	Wednesday, January 31, 2024	Chapter 3 Cont.	3.6-7	3-37C, 3-41E, 3-48, 3-58E, 3-69E, 3-70	<b>Quiz 2 Covering Ch. 3</b>
4	Monday, February 5, 2024	Test Review 1 Chapters 1-3	-	-	-
	Wednesday, February 7, 2024	-	-	-	<b>TEST 1 Chapters 1 – 3</b>
5	Monday, February 12, 2024	4 – Properties of Pure Substances	4.1-3	Understand T-v, P-v, and P-T Diagrams	-
	Wednesday, February 14, 2024	Chapter 4 Cont.	4.4-5	4-6C, 4-7C, 4-10C, 4-15C, 4-26, 4-27E	<b>Quiz 3 Covering Chap. 4</b>



6	Monday, February 19, 2024	4 – Properties of Pure Substances	4.5	4-35E, 4-36E, 4-39, 4-43, 4-46, 4-49, 4-50, 4-54E, 4-59,	-
	Wednesday, February 21, 2024	Chapter 4 Cont.	4.6	4-62, 4-64, 4-71, 4-74E	<b>Quiz 4 Covering Chap. 4</b>
7	Monday, February 26, 2024	5 – Energy Analysis of Closed Systems	5.1-3	5-3, 5-8, 5-9, 5-12, 5-26E, 5-29E, 5-32, 5-33, 5-34	<b>Quiz 5 Covering Chap. 4</b>
	Wednesday, February 28, 2024	5 – Energy Analysis of Closed Systems	5.4-5	5- 50E, 5-56, 5-67E, 5-68, 5-69, 5-78, 5-79, 5-85, 5-86	-
8	Monday, March 4, 2024	6- Mass and Energy Analysis of Control Volumes	6.1-2	6-6E, 6-17, 6-18E, 6-20E, 6-28, 6-39E	<b>Quiz 6 Covering Chap. 5</b>
	Wednesday, March 6, 2024	Test Review 2 Chapters 4 and 5	-	-	-
9	Monday, March 11, 2024 Wednesday, March 13, 2024		Spring Break		
	Monday, March 18, 2024	-	-	-	<b>TEST 2 Chaps. 4 and 5</b>
10	Wednesday, March 20, 2024	6- Mass and Energy Analysis of Control Volumes	6.3-4	6-44, 6-46E, 6-60E, 6-71, 6-75E, 6-78, 6-93E, 6-107, 6-108	-
11	Monday, March 25, 2024	Finish Chapter 6	-	-	<b>Quiz 7 Covering Ch. 6</b>
	Wednesday, March 27, 2024	Chapter 7 – Second Law of Thermodynamics	7.1-4	7-17E, 7-20, 7-21, 7-22, 7-42, 7-	-

				43, 7-77, 7-78, 7-79, 7-90,	
12	Monday, April 1, 2024	Chapter 7 Cont.	7.5-10	7-91, 7-94E, 7-96	<b>Quiz 8 Covering Chap. 7</b>
	Wednesday, April 3, 2024	Finish Chapter 7	7.5-10	-	-
13	Monday, April 8, 2024	Part 2 Chapter 11 – Introduction and Properties of Fluids	11.1-2	11-1C, 11-3C, 11-4C, 11-5C, 11-6C, 11-7C	<b>Quiz 9 Covering Chap. 7</b>
	Wednesday, April 10, 2024	Test Review 3 Chapters 6, 7, and 11	-	-	-
14	Monday, April 15, 2024	-	-	-	<b>TEST 3 Chaps. 6, 7 and 11</b>
	Wednesday, April 17, 2024	Chapter 11 – Fluid Statics	11.3-4	11-8, 11-10, 11-14, 11-16	-
15	Monday, April 22, 2024	Chapter 12 – Bernoulli and Energy Equations	12.1-3	12-16, 12-17, 12-26, 12-31, 12-32, 12-35E	<b>Quiz 10 Covering Chap. 11</b>
	Wednesday, April 24, 2024	Chapter 12 Cont.	12.1-3	-	-
16	Monday, April 29, 2024	Finish Chapter 12	-	12-48, 12-50, 12-51, 12-52, 12-54, 12-55, 12-60, 12-64E	-
	Wednesday, May 1, 2024	Chapter 13 – Momentum Analysis of Flow Systems	13.1-4	13-19, 13-20, 13-25, 13-32E	-
17	Wednesday, May 8, 2024	<b>FINAL EXAM: 4-5.30pm (TEST 4) Chap 11,12,13</b>			